

Morning Assignment 02 jun 2026
Chemical Kinetics Module-kin.02

1. What is meant by rate of a chemical reaction?
2. Why is rusting of iron considered a slow reaction?
3. Why is precipitation of silver chloride called an instantaneous reaction?
4. What is average rate of reaction?
5. What is instantaneous rate of reaction?
6. Why is a negative sign used in the rate expression for reactants?
7. What are the units of reaction rate?
8. How is instantaneous rate determined graphically?
9. Why are stoichiometric coefficients used in rate expressions?
10. How can the rate of a gaseous reaction be expressed?

[Help →Rate of reaction](#)